Tetrahedron Letters,Vol.25,No.44,pp 5043-5046,1984 0040-4039/84 \$3.00 + .00 Printed in Great Britain ©1984 Pergamon Press Ltd.

## VICINAL ALKYLATION OF ALKYNES. A SHORT ROUTE TOWARD $\Delta^{\alpha,\beta}$ butenolides, furans and cyclopentenones.

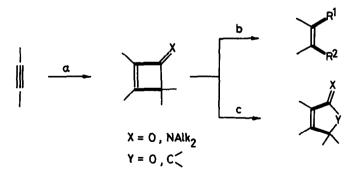
C. Schmit, S. Sahraoui-Taleb, E. Differding, C.G. Dehasse-De Lombaert and L. Ghosez

Laboratoire de Chimie Organique de Synthèse Université Catholique de Louvain, Place L. Pasteur, 1 B-1348 LOUVAIN-LA-NEUVE, BELGIUM.

Summary : Cyclobutenones 1 which are readily prepared from alkynes and keteniminium salts 2 were regiospecifically converted into  $\Delta \alpha_{\gamma,\beta}$  butenolides 4 or cyclopentenones 7. Reaction of 4 with dissobutylaluminum hydride yielded the corresponding substituted furans.

A methodology leading to the regio- and stereoselective addition of carbon chains to an acetylenic bond should be extremely useful. We have considered that this vicinal alkylation could possibly be achieved by a two-step sequence (Scheme 1) involving a [2+2] cycloaddition of a ketene or a keteniminium salt to an alkyne (step a) followed by cleavage of the  $C^1-C^4$  bond (step b) or, alternatively, the regiospecific insertion of an atom Y in that bond (step c).

Scheme 1. : Vicinal Alkylation of Acetylenes

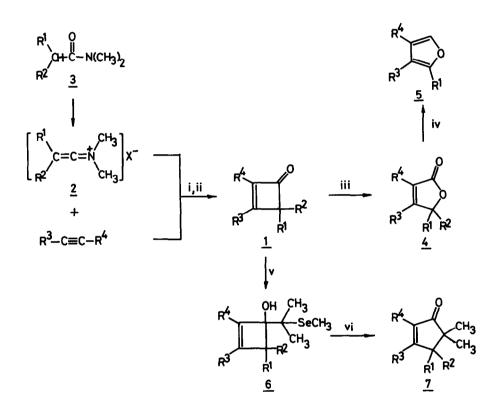


A demonstration of the utility of sequence a-c is the subject of this communication.

Previous studies<sup>1</sup> have shown that unactivated alkynes are rather poor ketenophiles. On the other hand we have shown that tetramethylketeniminium salts readily cycloadded to alkyl- and arylacetylenes<sup>2</sup>. This reaction has now been developed into an efficient and general method of synthesis of cyclobutenones 1.

The results are shown in Scheme 2 and Table 1. When  $R^{1}=R^{2}=CH_{3}$ , the keteniminium salt 2 was generated from the readily available <sup>3</sup> tetramethyl- $\alpha$ -chloroenamine and TiCl<sub>4</sub>. When  $R^{1}$  or (and)  $R^{2}$  are H, 2 was formed in situ from the appropriate tertiary amide 3, trifluoromethanesulfonic anhydride and collidine in refluxing dichloromethane <sup>4</sup>. Hydrolysis of the crude adducts gave good yields of cyclobutenones 1. Yields were good even in the case of the more sensitive trimethylsilylacetylene.

Scheme 2.



 $\begin{array}{l} \underline{\text{Experimental Conditions}} : \text{ i. } \text{CH}_2\text{Cl}_2, -40\,^\circ\text{C} (X=\text{TiCl}_5)^2 \text{ or reflux } (X=\text{CF}_3\text{SO}_3)^4; \text{ ii. } 0,1 \text{ N} \\ \hline \text{NaOH-CH}_2\text{Cl}_2; \text{ iii. } \text{m CPBA } (1.6 \text{ equiv.}), \text{ NaHCO}_3 (6.6 \text{ equiv.}) \text{ in } \text{CH}_2\text{Cl}_2; \text{ iv. Dibal } (1.5 \text{ equiv.}), \\ \hline \text{THF}, - 20\,^\circ\text{C}; \text{ v. } (\text{CH}_3)_2\text{C}(\text{SeCH}_3)\text{Li } (1 \text{ equiv.}) \text{ in } (\text{C}_2\text{H}_5)_2\text{O}, -78\,^\circ\text{C}, \text{ then } \text{NH}_4\text{Cl}^7; \text{ vi. } \text{KOH } 50\% \text{ -} \\ \hline \text{CHCl}_3 - \text{C}_6\text{H}_5\text{CH}_2\text{NEt}_3\text{Cl} (\text{entry } 7\text{g}) \text{ or } \text{TlOC}_2\text{H}_5 \text{ in } \text{CHCl}_3 (\text{entry } 7\text{h})^7. \end{array}$ 

Table 1 : Yields of Compounds 1, 4, 5, 6 and 7

Entry	R <sup>1</sup>	$R^2$	r <sup>3</sup>	R <sup>4</sup>	Yields <sup>a</sup> , (%)				
					<u>1</u>	<u>4</u>	<u>5</u>	<u>6</u>	<u>7</u>
a	снз	снз	с <sub>6</sub> н <sub>5</sub>	с <sub>6</sub> н <sub>5</sub>	82 <sup>b</sup>	57 <sup>d</sup>			
b	сн3	CH3	снз	CH3	70 <sup>b</sup>	51 <sup>e</sup>			
с	снз	снз	(CH <sub>3</sub> ) <sub>3</sub> C	H	61 <sup>b</sup>	66 <sup>e</sup>			
d	снз	Н	с <sub>б</sub> н <sub>5</sub>	H	70 <sup>C</sup>	52 <sup>d</sup>			
е	снз	Н	(CH <sub>3</sub> ) <sub>3</sub> C	Н	69 <sup>C</sup>	61 <sup>e</sup>			
f	Н	Н	(CH <sub>3</sub> ) <sub>3</sub> C	Н	32 <sup>C</sup>	53 <sup>e</sup>	57 <sup>e</sup>		
g	снз	сн <sub>з</sub>	(CH <sub>3</sub> ) <sub>3</sub> S1	H	88 <sup>b</sup>	76 <sup>e</sup>	77 <sup>e</sup>	44 <sup>e</sup>	58 <sup>e</sup>
h	снз	Н	(CH <sub>3</sub> ) <sub>3</sub> Si	н	89 <sup>c</sup>	98 <sup>e</sup>		52 <sup>e</sup>	70 <sup>e</sup>

 $^a$  Yields refer to pure isolated compounds which were fully characterized by spectral data.  $^b$  From tetramethyl- $\alpha$ -chloroenamine  $^c$  From amides 3.

<sup>d</sup> Purified by recrystallisation

<sup>e</sup> Purified by chromatography on silicagel

The success of our strategy rested upon the regioselectivity of the ring-expansion reaction. Baeyer-Villiger oxidations of cyclohexenones and other  $\alpha,\beta$  unsaturated ketones normally occur with preferential migration of the vinyl group over the alkyl group<sup>5</sup>. However we anticipated that this sequence of migratory aptitudes could be reversed in cyclobutenones <u>1</u> as a result of the allylic character of C-4. Indeed the reaction of <u>1</u> with m-chloroperbenzoic acid exclusively yielded the  $\Delta^{\alpha,\beta}$  butenolides <u>4</u>. These were readily transformed by a known procedure<sup>6</sup> into the corresponding derivatives of furan 5.

Finally, using Krief's method<sup>7</sup> for the ring enlargement of cyclic ketones, we have been able to convert <u>1</u> into cyclopentenones <u>7</u> via the intermediate  $\beta$ -hydroxyselenide <u>6</u>. The ring expansion also occured with preferential migration of the allylic  $C^1-C^4$  bond over the  $C^1-C^2$  vinylic bond.

Thus a two-step method to effect the vicinal alkylation of alkynes is now available. The mild conditions required for the reactions, the total regioselectivity, the ease of operation and good yields combine to make this a practical method of synthesis of  $\Delta^{\alpha,\beta}$  butenolides or substituted furans or cyclopentenones.

## Acknowledgments

We thank Professor A. Krief (Fac. N.D. de la Paix, Namur, Belgium) for helping us in performing the cyclopentenone synthesis. We thank Ciba-Geigy, Basel, The "Service de la Programmation de la Politique Scientifique" (Grant 79/84-13) and the "Administration Générale de la Coopération au Développement" for financial support.

## References

- L. Ghosez and M.J. O'Donnell, "Pericyclic Reactions", A.P. Marchand, R.E. Lehr, Eds, Academic Press, New York, 1977, Vol.II, Chapter 2 p. 79
- 2. C. Hoornaert, A.M. Frisque and L. Ghosez, Angew. Chem. Int. Ed. 1975, 14, 569.
- B. Haveaux, A. Dekoker, M. Rens, A.R. Sidani, J. Toye and L. Ghosez, Org. Synth., 1980, 59, 26.
- J.B. Falmagne, J. Escudero, S. Sahraoui-Taleb and L. Ghosez, Angew. Chem. Int. Ed., 1981, 20, 926.
- 5. See, for example : E. Wenkert and M. Rubin, Nature (London) (1952), 170, 708 ; J.R. Handley, A.A. Swigar, R.M. Silverstein, J. Org. Chem. 1979, 44, 2954; G.A. Krafft, J.A. Katzenellenbogen, J. Am. Chem. Soc. 1981, 103, 5459 ; H. Tomioka, K. Oshima and H. Nozaki, Tetrahedron Lett. 1982, 23, 99.
- 6. H. Minato and T. Nagasaki, Chem. Ind. (London) 1965, 899 and J. Chem. Soc., 1966, 377.
- D. Labar, J.L. Laboureur, A. Krief, Tetrahedron Lett. 1982, 983; J.L. Laboureur and A. Krief, ibid., 1984, 2713.

(Received in France 20 August 1984)